

D. S. COOMBS, A. J. ELLIS, W. S. FYFE and A. M. TAYLOR

Li_2O tr.; $\text{H}_2\text{O} + 110^\circ 12\cdot45$; $\text{H}_2\text{O} - 110^\circ 3\cdot3$; TiO_2 n.f.; P_2O_5 0·01; MnO n.f., CO_2 n.f.
Spectrographic analysis: Ag 0·015; Pb 0·015; Sn <0·01; Ba 0·05. Not detected: V, Cd, As, Be, Zn, W, Mo, Ni, Cr, Co, Bi, Sb, B, La, Yt, Ce. Total: 99·44. J. A. RITCHIE anal., s. g. 2·23 ± 0·01.

Structural formula: $(\text{Ca}_{2\cdot9}\text{Sr}_{0\cdot5}\text{Na}_{1\cdot5}\text{K}_{0\cdot5})\text{Al}_{0\cdot2}\text{Si}_{26\cdot8}\text{O}_{72\cdot24}\cdot8\text{H}_2\text{O}$ in satisfactory agreement with that of HEY and BANNISTER (1934) and Appendix C above.

Refractive indices variable, $\alpha 1\cdot500-1\cdot505 \pm 0\cdot001$, $\beta 1\cdot500-1\cdot506 \pm 0\cdot001$, $\gamma 1\cdot506-1\cdot512 \pm 0\cdot002$.

Prehnite

Locality: Prospect Quarry, New South Wales. Large, translucent, very pale green botryoidal masses in cavities in dolerite. Average of comparable analyses by A.M.T. and by J. A. RITCHIE, Dominion Laboratory, Wellington: SiO_2 43·7; Al_2O_3 24·05; Fe_2O_3 0·93; FeO 0·03; MgO 0·11; CaO 26·85; Na_2O 0·04; K_2O n.f.; Li_2O n.f.; $\text{H}_2\text{O} + 110^\circ 4\cdot54$; $\text{H}_2\text{O} - 110^\circ 0\cdot03$; TiO_2 tr.; P_2O_5 0·02; MnO n.f.; CO_2 n.f.; Spectrographic analysis (Dominion Laboratory): Ag 0·01; Pb 0·04; SnO 0·01; BaO <0·003. Not detected: V, Cd, etc., as above. Total: 100·36; s. g. 2·93.

Formula on basis of eleven oxygen atoms:

$(\text{Ca}_{1\cdot98}\text{Mg}_{0\cdot01}\text{Na}_{0\cdot01})(\text{Al}_{1\cdot95}\text{Fe}_{0\cdot05})\text{Si}_{3\cdot01}\text{O}_{11\cdot1}\cdot04\text{H}_2\text{O}$ in very close agreement with the conventional formula $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_{10}(\text{OH})_2$. $\alpha 1\cdot615 \pm 0\cdot001$; $\beta 1\cdot624 \pm 0\cdot001$; $\gamma 1\cdot643 \pm 0\cdot002$; $\gamma - \alpha 0\cdot028$; $2V_\gamma 69^\circ \pm 2^\circ r > v$ strong.

As is common in prehnite from such environments, and in contrast to the prehnite described in Section 2 above from the New Zealand greywackes, basal sections show complex lamellar twinning, and when thick, anomalous interference tints and incomplete extinction.

REFERENCES

- AMES L. L. and SAND L. B. (1958) Hydrothermal synthesis of wairakite and calcium-mordenite. *Amer. Min.* **43**, 476-480.
- AMES A. C. (1950) An intrusion of porphyrite near Waihao Forks, South Canterbury. *Trans. Roy. Soc. N.Z.* **78**, 271-279.
- AMES A. C. (1952) Petrological features of the rocks of the Maruwenau district, North Otago. *Trans. Roy. Soc. N.Z.* **79**, 376-385.
- ANTUN P. (1953) Laumontite de Serpent. *Ann. Soc. Géol. Belg.* **77**, B63-71.
- BANWELL C. J., COOPER E. R., THOMPSON G. E. K. and MCCREE K. J. (1957) Physics of the New Zealand thermal area. *Bull. N.Z. Dep. Sci. Industr. Res.*
- BARRIER R. M. (1948) Syntheses and reactions of mordenite. *J. Chem. Soc.* 2158-2163.
- BARRIER R. M. (1950) Ion-exchange and ion-sieve processes in crystalline zeolites. *J. Chem. Soc.* 2342-2350.
- BARRIER R. M. and WHITE E. A. D. (1952) Synthetic crystalline sodium aluminosilicates. *J. Chem. Soc.* 1561-1571.
- BEATTIE I. R. (1954) The structure of analcite and ion-exchanged forms of analcite. *Acta Cryst.* **7**, 357-359.
- BENSON W. N. (1942) The basic igneous rocks of Eastern Otago and their tectonic environment, Part 3. *Trans. Roy. Soc. N.Z.* **72**, 160-185.
- BOLDYREVA A. M. (1953) Authigenic analcite in Upper Permian sediments of the Chkalova and Aktyubinsk districts. *Mém. soc. russe min.* **82**, 291-297; *Chem. Abstr.* **48**, 5030f.
- BRADLEY W. H. (1928) Zeolite beds in the Green River formation. *Science* **67**, 73-74.
- BRADLEY W. H. (1929) The occurrence and origin of analcite and meerschaum beds in the Green River formation of Utah, Colorado and Wyoming. *U.S. Geol. Surv. Prof. Paper A* **158**, 1-7.
- BRAMLETT M. N. and POSNJAK E. (1933) Zeolitic alteration of pyroclastics. *Amer. Min.* **18**, 167-171.
- BRODERICK T. M. (1929) Zoning in Michigan copper deposits and its significance. *Econ. Geol.* **24**, 149-162 and 311-326.